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## Section 16 2 Thermal Energy And Matter Answers

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Thermodynamics. States that absolute zero cannot be reached.

Section 16.2 - Heat and Thermodynamics Flashcards | Quizlet

- Heat is a measure of energy (Joules)
  - Transfer of thermal energy from one object to the next because of temperature difference.
  - Heat flows naturally from hot objects to cold objects.
  - HEAT and TEMPERATURE are NOT the same thing.
  - Ex. A cold Lake Superior has more heat energy than a boiling pot of water.
- Heat

Section 16.2 Thermal Energy

Recall that thermal energy is the total potential and kinetic energy of all the particles in an object. Thermal energy depends on the mass, temperature, and phase (solid, liquid, or gas) of an object. Thermal energy, unlike temperature, depends on mass. Suppose you compare a cup of tea and a teapot full of tea.

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