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Both compounds and mixtures contain more than one kind of atom. Explain how a compound is different from a mixture. A compound is a substance that has more than one type of substance in fixed, specific ratios. A mixture contains more than one substance, which can either be elements or compounds in no fixed ratio.

Compounds and Mixtures Section 3

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•An element is matter that is composed of one type of atom, which is the smallest piece of matter that still retains the property of the element. •Atoms are composed of particles called protons, neutrons, and electrons. Atomic Components Structure of the Atom. Click image to view movie.

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3. Directions: Compare static friction, sliding friction, rolling friction, and air resistance in the chart. Directions: Use the terms from the word bank to fill in the blanks in front of the correct phrases below.

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?? 3. Make a table that lists the substances and at least one element in the substance, or fill in the table on the next page. ?? Make a physical change to each substance and observe. ?? 4. In your data table, record the physical change.

The Periodic Table and Physical Properties; Grade 8 Chapter 7

Directions: Use the terms from the word

bank to fill in the blanks in the summary statements. DNA gender sunlight environment hormones temperature genes selecting traits. Eye, hair, and skin color you inherit from your parents are called inherited (1) _____.

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Chemical Families Section 6-3.... KEY CONCEPTS. - ~ Elements within the same family of the periodic table have similar properties because they have the same number of valence electrons. M Vocabulary Skills: Applying Definitions On the periodic table below, label each of the following: acnoid series alkaline earth ...

REVIEW and REINFORCEMENT Section 6-3

3. any element in Group 2 4. any element in Group 18 5. any element in Group 16 Answer the following questions in the spaces provided. use the periodic table if you need help. 1. What is the usual oxidation number of oxygen? 2. What is the usual oxidation number of hydrogen? 3.

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Compounds and Mixtures Section 3
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- An element is matter that is composed of one type of atom, which is the smallest piece of matter that still retains the property of the element.
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Chemical Families Section 6-3.... KEY
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3. Directions: Compare static friction, sliding friction, rolling friction, and air resistance in the chart. Directions: Use the terms from the word bank to fill in the blanks in front of the correct phrases below.

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Both compounds and mixtures
contain more than one kind of atom.
Explain how a compound is different
from a mixture. A compound is a
substance that has more than one
type of substance in fixed, specific
ratios. A mixture contains more
than one substance, which can
either be elements or compounds in
no fixed ratio.

Compounds and Mixtures Section 3
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